

Peng–Robinson–Stryjek–Vera (PRSV) EOS @model

@wikipedia

One of the cubic equations of real gas state defining the Compressibility factor $Z(p, T)$ as a function of fluid pressure p and fluid temperature T :

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|--|---|
| (1) $Z^3 - (1 - B)Z^2 + (A - 2B - 3B^2)Z - (AB - B^2 - B^3) = 0$ | |
| (2) $A = 0.45724 \cdot \alpha \cdot \frac{p_r}{T_r^2}$ | (3) $B = 0.07780 \cdot \frac{p_r}{T_r}$ |
| (4) $\alpha = (1 + \kappa(1 - T_r^{0.5}))^2$ | (5) $\kappa = \kappa_0 + [\kappa_1 + \kappa_2(\kappa_3 - T_r)(1 - T_r^{0.5})] (1 + T_r^{0.5})(0.7 - T_r)$ |
| | (6) $\kappa_0 = 0.378893 + 1.4897153 \omega - 0.17131848 \omega^2 + 0.0196554 \omega^3$ |

where

| | | | |
|----------|------------------------|------------------------------------|----------------------|
| Z | Compressibility factor | p_c | Critical pressure |
| p | Fluid pressure | T_c | Critical temperature |
| T | Fluid temperature | $p_r = p/p_c$ | Reduced pressure |
| R | Gas constant | $T_r = T/T_c$ | Reduced temperature |
| ω | Acentric factor | $\{\kappa_1, \kappa_2, \kappa_3\}$ | fitting parameters |

Once compressibility Z-factor $Z(p, T)$ is known the fluid density ρ can be calculated as:

$$(7) \quad \rho(p, T) = \frac{1}{Z(p, T)} \cdot \frac{M}{R} \cdot \frac{p}{T}$$

where

| | |
|-----|------------------|
| M | fluid molar mass |
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See also

Natural Science / Physics / Thermodynamics / Equation of State / Real Gas EOS @model

[Real Gas]

