

Mixture

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A [chemical substance](#) consisting of two or more pure substances (called [mixture components](#) or [mixture ingredient](#)) with chemical bonding between [components](#) being weaker than that of between molecules of [components](#).

A list of all [chemical substances](#) constituting a [mixture](#) is called [Mixture composition](#).

The chemical properties of [mixtures](#) may differ from those of any constituent [components](#).

Unlike [pure substances](#) the [mixtures](#) in a state of [Thermodynamic equilibrium](#) may see various [phases](#) occupy a continuous area in [PT Diagram](#) (see for example [Vapour Liquid Equilibrium](#)).

Mixtures with two [components](#) are called [Binary Mixtures](#) and apart from a wide practical value they have a zoomed value in academic studies which help understand and model behaviour of all [mixtures](#).

A mixture in a [fluid state](#) (liquid, gas or supercritical fluid) is called a [fluid mixture](#).

In [petroleum industry](#) the most popular [mixtures](#) are [hydrocarbons](#) and [water](#) (H_2O).

See Also

[Natural Science / Chemistry / Chemical Substance](#)

[\[Pure substance \]](#) [\[Fluid Mixture \]](#)

[\[Mixture composition \]](#)

[\[Thermodynamic equilibrium \]](#) [\[Vapour Liquid Equilibrium \(VLE\) \]](#)